## Lignans from the Bark of Zanthoxylum planispinum

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A new dimeric lignan, named biplanispine A (1) which represents the first example of a dimeric 2,6 diaryl-3,7-dioxabicyclo[3.3.0]octane skeleton from a Rutaceae plant, together with 13 known lignans 2-14, was isolated from the bark of Zanthoxylum planispinum. Their structures were elucidated by spectroscopic methods, especially 2D-NMR techniques. All isolated compounds were evaluated for their antibacterial activities by the disc diffusion method. Compounds 7 and 9 showed interesting antibacterial activities at 200 µg/disc; they were as effective as kanamycin sulfate at 10 µg/disc against Staphylococcus aureus and Bacillus subtilis.

Introduction. – The genus Zanthoxylum belongs to the Rutaceae family, which comprise 250 species distributed in tropical and subtropical zones of Asia, America, Africa, and Oceania, and there are 45 species and 13 mutations in China [1]. Chemical constituents of Zanthoxylum have been studied extensively. Previous phytochemical investigations on this species revealed that benzophenanthridine alkaloids [2], coumarins [3], amides [4], and lignans [5] are largely represented in this genus. Lignan constituents have been reported to possess anticancer, antimicrobial, antioxidant, anti-inflammatory, and immunosuppressive activities [6]. Zanthoxylum planispinum is widely used as a folk medicine for treating colds, preventing toothache, and expelling roundworms [7]. Chemical studies of the bark of Z. planispinum have not been conducted previously. In our search for bioactive compounds from the crude extract of ethnomedicinal plants, we found that an AcOEt-soluble partition of the EtOH extract of the bark of Z. planispinum exhibited potent antibacterial activities against some microbes. This prompted us to perform a detailed bioassay-guided investigation of this plant. As a result, a new dimeric lignan, named biplanispine  $A(1)$ , which is the first example of dimeric 2,6-diaryl-3,7-dioxabicyclo[3.3.0]octane skeleton from Rutaceae plant, together with 13 known lignans  $2-14$ , were isolated from the bark of Z. planispinum. Their structures were elucidated by spectroscopic methods, especially 2D-NMR techniques. This article describes the structural investigation of these natural products and their antibacterial activities.

Results and Discussion. – Compound 1, which was obtained as colorless amorphous powder, exhibited an  $[M + Na]$ <sup>+</sup> ion peak at  $m/z$  765.2140 in the HR-ESI-MS (positiveion mode), corresponding to the molecular formular  $C_{40}H_{38}O_{14}$ . The <sup>1</sup>H- and <sup>13</sup>C-NMR

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data (Tables 1 and 2) suggested that 1 possesses two 2,6-diaryl-3,7-dioxabicyclo[3.3.0]octane units. <sup>1</sup>H,<sup>1</sup>H-COSY, HMBC (*Tables 1* and 2) and ROESY experiments, as well as comparison with known compounds established the structure of 1 as shown in the Figure, and was named biplanispine A.

The  ${}^{1}H$ - and  ${}^{13}C$ -NMR data revealed that part **A** of 1 resembled those of the known compound  $3$  [8]. The <sup>1</sup>H-NMR spectrum showed two CH groups at  $\delta$ (H) 2.47 – 2.49 (*m*, 1 H) and at  $3.38 - 3.42$  (*m*, 1 H), two benzylic O-bearing CH moieties at  $\delta$ (H) 4.34 (*d*,  $J = 7.2$ , 1 H) and 4.78 (d,  $J = 6.0$ , 1 H), two O-bearing CH<sub>2</sub> groups at  $\delta$ (H) 4.05 (d, J = 9.0, 1 H) and 3.68 – 3.76 (m, 1 H), 2.88 (dd,  $J = 8.4, 7.8, 1$  H), and 2.99 (dd,  $J = 9.0, 7.8$ , 1 H), six aromatic H-atoms as two  $ABX$  systems at  $\delta$ (H) 6.82 – 7.18 indicating the presence of two 1,3,4-trisubstituted benzene rings, in addition to one MeO group at  $\delta(H)$  3.63 (s, 3 H), and of one OCH<sub>2</sub>O group at  $\delta(H)$  5.96 (s, 2 H). Accordingly, part **A** of 1 was assigned to be a 2,6-diaryl-3,7-dioxabicyclo[3.3.0]octane bearing one MeO

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	$\delta(H)$	$\delta(C)$	HMBC $(H \rightarrow C)$
$H-C(1)$	$2.47 - 2.49$ ( <i>m</i> )	54.0	C(9)
$H-C(2)$	4.34 $(d, J = 7.2)$	86.7	C(1), C(8), C(9), C(10), C(14)
CH <sub>2</sub> (4)	2.88 (dd, $J = 8.4, 7.8$ ), 2.99 (dd, $J = 9.0, 7.8$ )	68.9	
$H-C(5)$	$3.38 - 3.42$ ( <i>m</i> )	49.2	
$H-C(6)$	4.78 $(d, J=6.0)$	81.1	$C(4)$ , $C(5)$ , $C(15)$ , $C(16)$ , $C(20)$
CH <sub>2</sub> (8)	$3.68 - 3.76$ ( <i>m</i> ), $4.05$ ( <i>d</i> , $J = 9.0$ )	70.3	C(2), C(5), C(6)
C(9)		135.5	
$H - C(10)$	6.89 (br. $s$ )	106.5	C(2), C(11), C(12), C(14)
C(11)		147.4	
C(12)		146.6	
$H - C(13)$	$6.82 - 6.86$ $(m)$	108.0	
$H - C(14)$	$6.82 - 6.86$ ( <i>m</i> )	119.4	
C(15)		136.6	
$H - C(16)$	6.94 (br. $s$ )	109.9	$C(6)$ , $C(17)$ , $C(18)$ , $C(20)$
C(17)		152.0	
C(18)		140.8	
$H - C(19)$	$6.82 - 6.86$ ( <i>m</i> )	117.5	
$H - C(20)$	7.18 $(d, J = 8.4)$	124.4	$C(15)$ , $C(17)$ , $C(18)$
$-OCH2O-$	5.96 $(s)$	100.9	C(11), C(12)
MeO	3.63(s)	55.3	C(17)

Table 1. <sup>T</sup>H- and <sup>13</sup>C-NMR, and HMBC Data ((D<sub>6</sub>)DMSO) of Part **A** of Compound **1**.  $\delta$  in ppm, J in Hz.

Table 2. <sup>*I*</sup>H- and <sup>13</sup>C-NMR, and HMBC Data ((D<sub>6</sub>)DMSO) of Part **B** of Compound **1**.  $\delta$  in ppm, J in Hz.

	$\delta(H)$	$\delta(C)$	HMBC $(H \rightarrow C)$
$H - C(1')$	$3.74 - 3.76$ ( <i>m</i> )	48.4	C(8')
$H-C(2')$	$3.92 - 3.96$ ( <i>m</i> )	83.8	
CH <sub>2</sub> (4')	3.48 (dd, $J = 8.4, 7.8$ ), 3.66 – 3.74 (m)	69.1	C(1'), C(2')
$H-C(5')$	$3.26 - 3.32$ ( <i>m</i> )	49.4	
$H-C(6')$	4.73 $(d, J = 5.4)$	80.9	$C(4')$ , $C(5')$ , $C(15')$ , $C(16')$ , $C(20')$
CH <sub>2</sub> (8')	$3.68 - 3.76$ ( <i>m</i> ), 3.85 ( <i>d</i> , <i>J</i> = 8.4)	70.8	$C(2')$ , $C(5')$ , $C(6')$
$H-C(9')$	2.92 $(d, J = 10.2)$	50.6	$C(2')$ , $C(10')$ , $C(14')$
$H - C(10')$	$3.96 - 4.00(m)$	72.1	C(14')
C(11')		105.3	
C(12')		167.0	
$H - C(13')$	5.41 $(s)$	102.0	$C(9')$ , $C(11')$ , $C(12')$
C(14')		196.4	
C(15')		132.8	
$H - C(16')$	$6.82 - 6.86$ ( <i>m</i> )	106.2	
C(17')		147.0	
C(18')		145.9	
$H - C(19')$	$6.82 - 6.86$ ( <i>m</i> )	107.9	
$H - C(20')$	6.77 $(d, J = 7.8)$	118.6	$C(16')$ , $C(18')$
$- OCH2O -$	5.96 $(s)$	100.8	$C(17')$ , $C(18')$
$-OCH2O-$	5.25 $(s)$ , 5.59 $(s)$	100.3	C(11'), C(12')
$HO-C(10')$	5.84 $(d, J = 7.2)$		$C(9')$ , $C(10')$



Figure. The structure and arbitrary atom numbering of compound 1

group, one O-functional group and one  $OCH<sub>2</sub>O$  group. This assumption was further supported by the <sup>13</sup>C-NMR data including two CH groups ( $\delta$ (C) 54.0 and 49.2), two benzylic O-bearing CH groups ( $\delta$ (C) 86.7 and 81.1), two O-bearing CH<sub>2</sub> groups ( $\delta$ (C) 70.3 and 68.9), six aromatic C-atoms and six aromatic CH groups. The <sup>1</sup> H- and <sup>13</sup>C-NMR data of part **A** of 1 were similar to those of 3 [8], suggesting one MeO group of 3 was replaced by an O-functional group in 1, which was supported by an MS fragment at  $m/z$  356. In the HMBC spectrum, the correlations of  $\delta(H)$  3.63 (s, MeO)/  $\delta$ (C) 152.0 (C(17)) indicated that a MeO group was located at C(17). Furthermore, the ROESY spectrum allowed us to confirm the position of the MeO group, showing that the MeO signal correlates with H–C(16). The relative configuration of part  ${\bf A}$  of  ${\bf 1}$  was determined on the basis of a ROESY experiment. The ROESY correlations between  $H_{\beta}-C(2)/H_{\beta}-C(8)$  and  $H_{\beta}-C(4)$  indicated that these H-atoms were all on the same face of the molecule and had  $\beta$  orientation, whereas those observed with  $H_a-C(6)$ /  $\rm H_{\it a}$ –C(8) and  $\rm H_{\it a}$ –C(5), and  $\rm H_{\it a}$ –C(1)/ $\rm H_{\it a}$ –C(5) established that these  $\rm H$ -atoms were on the other side of the molecule and had  $\alpha$  orientation. The linkage between C(1) and  $C(5)$  was confirmed to be *cis*, which is typical of naturally occuring bis-tetrahydrofuran lignans [9]. From the ROESY data, part  $\bf{A}$  of 1 was concluded to be an unsymmetrically substituted bis-tetrahydrofuran lignan, possessing an equatorial aromatic ring at  $C(2)$  and an axial aromatic ring at  $C(6)$ .

Except for the  ${}^{1}$ H- and  ${}^{13}$ C-NMR signals of part **A** mentioned above, signals of three H-atoms of an ABX system at  $\delta(H)$  6.77 – 6.86 (*m*), an OCH<sub>2</sub>O at  $\delta(H)$  5.96 (s, 2 H), together with three aromatic C-atoms ( $\delta$ (C) 132.8, 147.0, 145.9) and three aromatic CH groups  $(\delta(C)$  106.2, 107.9, 118.6) were attributed to a 3,4-methylenedioxyphenyl group from the HMBC spectrum. Signals of two O-bearing CH groups  $(\delta(H)/\delta(C)$  3.92 – 3.96  $(m, 1 H)/83.8$  (C(2')) and 4.73 (d, J = 5.4, 1 H)/80.9 (C(6'))), two O-bearing CH<sub>2</sub> groups  $(\delta(H)/\delta(C)$  3.48  $(dd, J=8.4, 7.8, 1 H)$ , 3.66 – 3.74  $(m, 1 H)/69.1$  (C(4')) and 3.68 – 3.76  $(m, 1 H)$ , 3.85  $(d, J = 8.4, 1 H)/70.8$  (C(8'))), and two CH groups  $(\delta(H)/\delta(C)$  3.74 – 3.76  $(m, 1 H)/48.4$  (C(1')) and 3.26 – 3.32  $(m, 1 H)/49.4$  (C(5'))) were also apparent in the <sup>1</sup>H- and <sup>13</sup>C-NMR spectra. By analysis of the <sup>1</sup>H,<sup>1</sup>H-COSY and HSQC spectra, the correlation of  $H - C(2')$  with  $H - C(1')$ , of  $H - C(1')$  with  $H - C(5')$ and  $CH_2(8')$ , and of  $H-C(5')$  with  $H-C(6')$  and  $CH_2(4')$  gave the partial structure  $-OCH-CH(CH<sub>2</sub>O)-CH(CH<sub>2</sub>O)-CHO-$  which was consistent with the HMBC spectrum. The HMBC between  $H - C(6')$  and  $C(4')$  and  $C(5')$ , between  $CH_2(4')$  and

 $C(1')$  and  $C(2')$ , and between  $CH<sub>2</sub>(8')$  and  $C(2')$ ,  $C(5')$ , and  $C(6')$  were used to establish the presence of the 3.7-dioxabicyclo[3.3.0] octane skeleton in part **B** of 1. In the HMBC spectrum, the correlations of H–C(6') at  $\delta(H)$  4.73 (d, J=5.4, 1 H) to  $\delta(C)$  132.8  $(C(15'))$ , 106.2  $(C(16'))$ , and 118.6  $(C(20'))$  suggested that the 3,4-methylenedioxyphenyl group should be located at  $C(6')$  position. The remaining structure parts were identified by  $^1\mathrm{H}\text{-}$  and  $^{13}\mathrm{C}\text{-}\mathrm{NMR}$  (DEPT) as two CH (one O-bearing), one OCH<sub>2</sub>O, one  $\alpha$ , $\beta$ -unsaturated keto group at  $\delta$ (C) 196.4, one doubly O-bearing cyclic ketal C-atom at  $\delta$ (C) 105.3, one C=C bond with signals at  $\delta$ (C) 102.0 (CH) and 167.0 (C), and one OH group.

The aforementioned groups accounted for 21 out of 22 degrees of unsaturation in compound 1, indicating the presence of an additional ring. From the <sup>1</sup>H,<sup>1</sup>H-COSY spectrum, the correlation of  $HO-C(10')$  at  $\delta(H)$  5.84  $(d,J=7.2,1 \text{ H})$  with  $H-C(10')$  at  $\delta$ (H) 3.96–4.00 (*m*, 1 H) and H–C(10') with H–C(9') at  $\delta$ (H) 2.92 (*d*, J=10.2, 1 H) led to the partial structure C(10′)–C(9′). The HMBC cross-peaks between the OCH<sub>2</sub>O group at  $\delta(H)$  5.25 (s, 1 H) and 5.59 (s, 1 H) and  $\delta(C)$  105.3 (C(11')) and  $\delta(C)$  167.0  $(C(12'))$  indicated that the OCH<sub>2</sub>O group was attached to  $C(11')$  and  $C(12')$ . The correlation of H-C(13') at  $\delta$ (H) 5.41 (s, 1 H) with C(11') and C(12') established the partial structure  $C(11') - C(14')$ . The HMBC correlations of  $H - C(13')$  with  $C(9')$  and of  $H-C(9')$  with  $C(10')$  and  $C(14')$  indicated that  $C(9')$  and  $C(10')$  were connected with  $C(14')$  and  $C(11')$ , respectively, and established the remaining structure as depicted as ring  $C$  (see Fig.). This ring is located at  $C(2')$  based on the HMBC correlation of  $H-C(9')$  with  $C(2')$ .

The relative configuration of part **B** of 1 was established on the basis of  ${}^{1}H\text{-}NMR$ data and a ROESY experiment. The chemical shift observed for  $H-C(6)$  at  $\delta(H)$  4.73  $(d, J = 5.4, 1 \text{ H})$  was in accordance with that reported where the 3,4-methylenedioxyphenyl group was equatorial [10], and  $H - C(6')$  and  $H - C(5')$  was trans-oriented, which was confirmed by the ROESY correlation between  $\mathrm{H{-}C(6')}$  and  $\mathrm{H}_{\beta}{-}\mathrm{C(8')}$  . The ROESY correlation between  $H - C(1')$  and  $H - C(9')$  indicated that the substituent at  $C(2')$  is syn to  $H - C(1')$ . The large coupling constant  $J = 10.2$  Hz between  $H - C(9')$  and  $H - C(10')$  suggested that  $H - C(9')$  and  $H - C(10')$  are arranged trans to each other [11]. The OCH<sub>2</sub>O group at  $\delta(H)$  5.25 (s, 1H) and 5.59 (s, 1H) showed no ROESY correlations with other H-atoms. Thus, the relative configuration at  $C(11')$  could not be determined from the ROESY spectrum. From these results, the partial structure B of 1 was confirmed. Since  $C(18)$  of part **A** and  $C(11')$  of part **B** were O-bearing quaternary C-atoms, compound 1 is suggested to be a dimeric lignan linked through the dehydrogenation between  $HO-C(18)$  and a hemiacetal system at  $C(11')$  to form an ether linkage. Therefore, biplanispine A was deduced to be as shown for 1. However, the absolute configuration of 1 was not determined.

Naturally occuring dilignans from Rutaceae plants are rare. There are some reports on dilignans such as arctignan F isolated from Arctium fructus and Wikstroemai indica [12]. However, biplanispine A (1) is the first example of a dilignan with two 3,7 dioxabicyclo[3.3.0]octane system from Rutaceae plant.

The 13 known compounds were identified as asarinin (2) [13], fargesin (3) [8], horsfieldin (4) [14], eudesmin (5) [15], epipinoresinol (6) [8], de-4'-O-methylyangambin (7) [16], lariciresinol dimethyl ether (8) [17], syringaresinol (9) [8], pinoresinol monomethyl ether  $(10)$  [8], yangambin  $(11)$  [18], 3'-O-demethylepipinoresinol  $(12)$  [19], pinoresinol (13) [8], and (1R,2S,5R,6S)-6-(4-hydroxy-3-methoxyphenyl)-2-(3,4 dihydroxyphenyl)-3,7-dioxabicyclo[3.3.0]octane (14) [20] by comparison of their spectroscopic data with those reported in the literature. Notably, all reported lignans were isolated from Z. planispinum for the first time.

All isolated compounds were evaluated for their antibacterial activities by the disc diffusion method [21]. The results are shown in *Table 3*. At 200  $\mu$ g/disc 9 showed 13 and 16 mm diameter of the inhibition zone against Staphylococcus aureus and Bacillus subtilis, respectively. The strain B. *subtilis* showed 13-mm inhibition diameter, and the strain S. aureus showed 10 mm inhibition diameter in the presence of 7. The other compounds showed no activities against the two bacteria strains at the same concentration.

Compound	Bacteria		
	S. aureus	<b>B.</b> subtilis	
	6 <sup>b</sup>	6	
$2^{\circ}$	6	6	
3	6	6	
	6	6	
5	6	h.	
6	6	6	
	10	13	
8	6	6	
9	13	16	
10	6	6	
11	6	6	
12	6	6	
13	6	6	
14	h	h	
kanamycin sulfate	19	19	

Table 3. Antibacterial Activities of Lignans Isolated from Z. planispinum<sup>a</sup>)

<sup>a</sup>) Values are means of three determinations. <sup>b</sup>) Diameter of the inhibition zone in mm, including the diameter of the filter paper disc (6 mm).  $\degree$ ) 200 µg Compound/disc, or 10 µg kanamycin sulfate/disc, respectively.

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## Experimental Part

General. TLC: Precoated silica gel  $GF_{254}$  plates (Qingdao Haiyang Chemical Co., Ltd., P. R. China). Column Chromatography (CC): silica gel (SiO<sub>2</sub>; 200-300 mesh; Qingdao Haiyang Chemical Co., Ltd., P. R. China) and  $C_{18}$  reversed-phase silica gel (YMC CO., LTD., Japan). HPLC: Ultimate 3000 HPLC system; Ultimate 3000 pump; Ultimate 3000 Variable Wavelength; column Waters  $5C_{18}$ -MS-II (10 × 250 mm). Optical rotation: *Perkin-Elmer 341* polarimeter. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra: *Bruker AM-300* and 600 instrument;  $\delta$  in ppm rel. to Me<sub>4</sub>Si as internal standard (=0 ppm), J in Hz. EI-MS and HR-EI-MS: Finnigan MAT-95 mass spectrometer (70 eV); in  $m/z$  (rel. %). ESI-MS and HR-ESI-MS: Finnigan LCQ-Deca and Waters/Micromass Q-Tof-Ultima mass spectrometers, resp., in m/z (rel. int).

Plant Material. The barks of Z. planispinum were collected from Badong County, Hubei Province, P. R. China and identified by Prof. *Dinrong Wan*, College of Pharmacy, South Central University for Nationalities. A voucher specimen was deposited with the Herbarium of College of Pharmacy, South Central University for Nationalities.

Extraction and Isolation. The milled, air-dried barks of Zanthoxylum planispinum (17 kg) were powdered and then extracted three times with MeOH at r.t., and the MeOH extract (1.6 kg) was suspended in 90% H<sub>2</sub>O/MeOH and then successively extracted with petroleum ether (PE), AcOEt, and BuOH. The AcOEt extract  $(47 g)$  was subjected to CC (SiO<sub>2</sub>; cyclohexane/Me<sub>2</sub>CO 95:5, 9:1, 8:2, 7:3, 1:1, and 0:1): Frs.  $1 - 8$ . Compounds  $2(100 \text{ mg})$  and  $3(2.8 \text{ g})$  were crystallized from Fr. 3 and Fr. 5, resp. Fr. 6 (8.2 g) was subjected to CC (SiO<sub>2</sub>, cyclohexane/AcOEt  $9:1 \rightarrow 0:1$ ): Frs. 6.1–6.6. Compound 4 (30 mg) was crystallized from Fr. 6.2. Fr. 6.3 (3.9 g) was subjected to CC (ODS; H<sub>2</sub>O/MeOH 7:3  $\rightarrow$  0:1): Frs. 6.3.1 – 6.3.7. Fr. 6.3.2 (1.3 g) was subjected to CC (SiO<sub>2</sub>, cyclohexane/AcOEt 8:2  $\rightarrow$  0:1): Frs. 6.3.2.1 – 6.3.2.4. Fr. 6.3.2.3 (73.9 mg) was further purified by CC (SiO<sub>2</sub>; CHCl<sub>3</sub>/MeOH 1:0  $\rightarrow$  0:1): 5 (30 mg). Fr. 6.4 (2.2 g) was subjected to CC (SiO<sub>2</sub>; CHCl<sub>3</sub>/AcOEt 1:0  $\rightarrow$  0:1): Frs. 6.4.1 – 6.4.8. Fr. 6.4.4 (187.4 mg) was purified by CC (SiO<sub>2</sub>; cyclohexane/AcOEt  $9:1 \rightarrow 0:1$ ) and followed by CC (SiO<sub>2</sub>; PE/ Me<sub>2</sub>CO 8 :  $2 \rightarrow 0$  : 1): 6 (16.4 mg). Fr. 6.5 (1.3 g) was subjected to CC (*ODS*; H<sub>2</sub>O/MeOH 7 :  $3 \rightarrow 0$  : 1): Frs. 6.5.1 – 6.5.8. Fr. 6.5.2 (138 mg) was subjected to CC (SiO<sub>2</sub>; CHCl<sub>3</sub>/MeOH 1:0  $\rightarrow$  0:1): 7 (16.8 mg) and  $8$  (25 mg). Compound  $1$  (20 mg) was crystallized from Fr. 6.5.6. Fr. 7 (2.1 g) was subjected to CC  $(SiO<sub>2</sub>; CHCl<sub>3</sub>/MeOH 1:0 \rightarrow 0:1): 9 (5.5 mg).$ 

The remaining AcOEt extract (307 g) was subjected to CC (SiO<sub>2</sub>; PE/Me<sub>2</sub>CO 9:1, 8:2, 7:3, 1:1, and 0 : 1): Frs.  $1 - 15$ . Fr. 9 (52.3 g) was subjected to CC (SiO<sub>2</sub>; cyclohexane/AcOEt  $9 : 1 \rightarrow 0 : 1$ ): Frs. 9.1 – 9.13. Fr. 9.8 (8.9 g) was subjected to CC (*ODS*; H<sub>2</sub>O/MeOH 7:3  $\rightarrow$  0:1) and then further purified by semiprep. HPLC (MeOH/H<sub>2</sub>O 60:40, 3 ml/min): **10** (44 mg) at  $t<sub>R</sub>$  15.1 min, **11** (10 mg) at 23.2 min. Fr. 9.10 (8.2 g) was subjected to CC (*ODS*; H<sub>2</sub>O/MeOH 7:3  $\rightarrow$  0:1): *Frs.* 9.10.1 – 9.10.12. Fr. 9.10.5 was further purified by semi-prep. HPLC (MeOH/H<sub>2</sub>O 60:40, 3 ml/min;  $t_R$  32.7 min): 12 (20 mg). Fr. 9.10.7 was further purified by CC (SiO<sub>2</sub>; cyclohexane/AcOEt  $9:1 \rightarrow 0:1$ ): **13** (5 mg). Fr. 10 (90 g) was subjected to CC (SiO<sub>2</sub>; CHCl<sub>3</sub>/Me<sub>2</sub>CO 98 :  $2 \rightarrow 0$  : 1): Frs. 10.1 – 10.7. Fr. 10.4 (3.9 g) was subjected to CC (ODS; H<sub>2</sub>O/ MeOH  $7:3 \rightarrow 0:1$ : Frs. 10.4.1 – 10.4.7. Fr. 10.4.3 (135 mg) was purified by CC (SiO<sub>2</sub>; cyclohexane/AcOEt  $9:1 \rightarrow 0:1$ : 14 (17.6 mg).

Biplanispine A  $=(-6R^*,7S^*)-6-[(3aS^*,4R^*,6aS^*)-4-(1,3-Benzodioxol-5-y)tetrahydro-1H,3H-flu-1]$ ro[3,4-c]furan-1-yl]-7a-{4-[ (1S\*,3aS\*,4R\*,6aS\*)-4-(1,3-benzodioxol-5-yl)tetrahydro-1H,3H-furo[3,4 c]furan-1-yl]-2-methoxyphenoxy}-7-hydroxy-7,7a-dihydro-1,3-benzodioxol-5(6H)-one; 1). Colorless amorphous powder.  $\lbrack a \rbrack_{D} = -9.3$  ( $c = 0.1$ , DMSO). <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Tables 1* and 2. EI-MS: 356  $(76)$ , 151  $(60)$ , 149  $(100)$ , 135  $(68)$ , 131  $(36)$ . HR-ESI-MS: 765.2140  $([M + Na]^+$ ,  $C_{40}H_{38}NaO_{14}^+$ ; calc. 765.2159).

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